

Yuji Okayasu, , Public Works Research Institute, Japan 土木研究所 岡安祐司

OCCURRENCE OF ESTROGEN-LIKE SUBSTANCES IN WASTEWATER IN JAPAN

Yuji Okayasu¹, Yutaka Suzuki¹, Koya Komori¹, Makoto Yasojima¹

¹Water Environmental Research Group, Public Works Research Institute, ¹1-6, Minamihara, Tsukuba-City, Ibaraki, 305-8516 ¹ okayasu@pwri.go.jp, htanaka@pwri.go.jp, komori@pwri.go.jp, yasoji44@pwri.go.jp

ABSTRACT

The field study was conducted at twenty WWTPs in Japan about following estrogens: estrone(E1); 17 β -estradiol(E2); 17 α -ethynylestradiol(EE2); estriol(E3); estrone-3-sulfate(E1-S); β -estradiol 3-sulfate (E2-S); estriol 3-sulfate(E3-S); estrone β -D-glucuronide(E1-G); β -estradiol 17-(β -D)-glucuronide(E2-G); estriol 3- (β -D)-glucuronide(E3-G); β -estradiol 3-sulfate 17-glucuronide(E2-S&G); and estradiol 3,17-disulfate(E2-diS). The median concentrations of the estrogens ranged from ND to as high as >100 ng/L. In the influent samples, the concentration of E1, E2 and E3 are the same levels as those were previously reported. The conjugated estrogens are higher than those of the free estrogens. The reduction of the free estrogens in WWTPs was good. The concentrations of a few conjugated estrogens (E1-S and E2-S) were declined in the WWTPs, while the levels of other conjugated estrogens (E3-S, E1-G, E2-G, E3-G, E2-S&G and E2-diS) were unchanged or increased in the WWTPs. Moreover, the other field study was conducted at twelve WWTPs in Japan about following substances: nonylphenol(NP); nonylphenol ethoxylates(NPEOs); and nonylphenoxy acetic acids(NPECs). The concentrations of NP and Long-EO-chain NPEOs were declined in WWTPs while the levels of short –EO-chain NPECs were increased in the WWTPs.

KEYWORDS

Endocrine disruptors, 17β -estradiol, estrone, estrogen, estrogen conjugates, nonylphenol, nonylphenol ethoxylate, nonylphenoxy acetic acid.

INTRODUCTION

In recent years a new problem has emerged in our water environment, namely, endocrine disruptors (EDs) that may adversely affect the reproductive functions of human beings and wildlife. In Japan the EDs issue has arisen since the book "OUR STOLEN FUTURE (Colborn et al., 1996)" was introduced in 1997. Contamination of water with EDs poses new and potential environmental (and social) problems. The Japan Environmental Agency (JEA) published strategic programs on environmental endocrine disruptors (SPEED '98), in which basic policies and specific approaches to the problem are documented (JEA, 1998). In this document, the JEA listed more than 70 chemicals that are suspected to cause abnormalities in animals at extremely low levels. The Ministry of Land, Infrastructure and Transport (MLIT) of Japan has decided to grasp EDs conditions in the water environment conducting extensive studies with major rivers and WWTPs (MLIT, 2001a). Among over 70 suspected substances, the MLIT selected 27 compounds for the river studies and 25 substances for the WWTPs studies, based on the annual production of the chemicals and the levels detected in the

environment. The MLIT particularly concerned female hormones originating from humans and animals. The study by the MLIT, thus far, found that estrogen represented by 17β-estradiol (E2) exists in river water and wastewater (including treated wastewater) at significant levels (MLIT, 2001a; Tanaka et al., 2001b, 2003).

Analytical methods currently available for EDs are limited their applications to certain chemicals. The method for the analysis of E2 in the early stage of the MLIT survey had been based on enzyme linked immunosorbent assay (ELISA), which can detect E2 as low as 0.2 ng/L. However, due to the potential "cross-reaction" problem, ELISA is limited in its applications to certain conditions when it is applied to domestic wastewaters. Recently, estrone (E1) has emerged as concerned EDs in water environment (MLIT, 2001b; Goda et al., 2001), and many other estrogen-like chemicals appear to have estrogenic effects on fish. Furthermore, naturally occurring estrogens (e.g., E1 and E2) and nonylphenol (NP) tend to have higher estrogenic potentials than other synthetic, industrial chemicals (Yakou et al., 1999; Tanaka et al., 2001b). Although E2 and 17α -ethynylestradiol (EE2) can be analyzed simultaneously using the GC/MS method (Huang et al., 2001), this method is rather cumbersome requiring a derivatization process. PWRI refined the analytical method developed by Komori et al. (2001) for the analysis of specific estrogens (i.e., E2, E1, and EE2) present in wastewater. This method uses a LC/MS/MS, but the derivatization process is not required. Estrogens are excreted by male as well as female animals. Prior to excretion, most estrogens are hydroxylated and conjugated to glucuronides, sulfates, and acetates. Because very few analytical methods (Ternes et al., 1999a; Belfroid et al., 1999) are capable of analyzing estrogenic compounds, relatively little work has been directed toward investigating impacts and occurrence of estrogens in water environment. PWRI refined an analytical procedure (Komori et al. 2003) that allows routine analysis of estrogens and their conjugates (i.e., glucuronides and sulfates conjugates) in wastewater based on the method by Komori *et al.* (2002),

On the other hand, nonylphenol (NP) is known to be byproduct of nonylphenol ethoxylates (NPEOs) which are used as detergent for industrial use (Ahel *et al.*, 1994). It is important that not only NP but also NPEOs and their related substances are analyzed when behavior of NP in wastewater treatment process is surveyed. NPEOs are biodegraded to shorter-EO-chain NPEOs or NPECs under aerobic condition, and then biodegraded to NP under anaerobic condition. NP is a suspected endocrine disrupting chemical. Moreover, shorter-EO-chain NPEO has higher toxicity than longer EO chain NPEO (Comber *et al.*, 1993). PWRI developed analytical methods (Yasojima et al., 2002a, 2002b) which can analyze NP, NPEO (EO chain length 1-15) and NPEC (EO chain length 1-10) in wastewater.

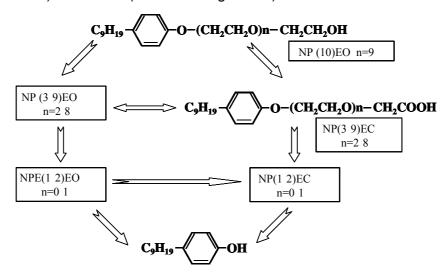


Figure 1 Degradation pathway of NP and its related substances

As mentioned above, result of YES (Yeast Estrogen Screening Test) shows that estrogens (E1, E2), EE2 and NP has high estrogenicity. This paper describes results of field study on estrogen, conjugated estrogens, NP and its related substances in wastewater treatment process.

METHOD

Field survey of estrogens and their conjugates

In this study, the analytical method by Komori et al. (2002) was refined for the analysis of estrogens and their conjugates in wastewater. Sample preparation of this method consists of solid-phase extraction with an Oasis HLB cartridge (for the filtrate), supersonic liquid extraction by methanol (for suspended matter), and cleaning with Sep-Pak Plus Florisil and Sep-Pak Plus NH₂. The pretreated (cleaned-up) sample was analyzed using a LC/MS/MS. A summary of the overall analytical scheme for WWTP influent and effluent are illustrated in Fig.2. First, a 500 ml wastewater sample was filtered through a 1-μm pore size glass fiber filter. Residue on the filter was extracted by supersonic extraction with 5 ml of methanol. The methanol extract was then added to the filtrate. A volume of 0.5 ml of 20% acetic acid, 2 ml of 0.5 mol/l ion pair coupling (IPC) solution and 40 ng of each internal standard [i.e., estrone-2,4-d2] 17β-estradiol-16,16,17-*d*3 (E1-d2).(E2-d3), 17α -ethynylestradiol-2,4,16,16-d4 (EE2-d4), estriol-2,4-d2 (E3-d2), and sodium 17 β -estradiol-2,4,16,16-d4 3-sulphate (E2-S-d4)] were added. After the mixing, the solution was passed through an Oasis HLB cartridge. Flow rate was maintained at 15 ml/min. The Oasis HLB cartridge was conditioned with methanol and purified water prior to extraction. The Oasis HLB cartridge was centrifuged with a gentle stream of nitrogen gas until it was dried completely. Then estrogen was eluted from the Oasis HLB cartridge with 6 ml of

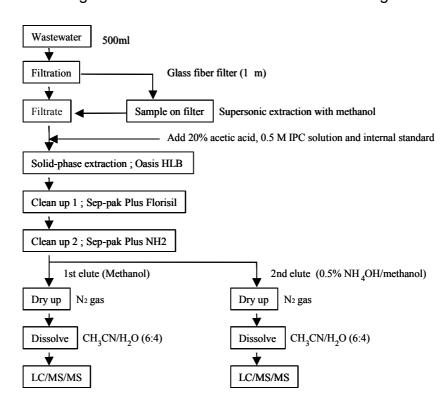


Figure 2 Flow diagram of analytical method

methanol. The eluent was blown down to dryness with a gentle stream of nitrogen gas. The dry residue was dissolved in 1 ml of hexane/dichloromethane (1:1, v/v) with supersonic extraction, and cleaned-up with Sep-Pak Plus Florisil. The cartridge was washed with 10 ml of hexane/dichloromethane (1:1, v/v). Estrogens were eluted from a Sep-Pak Plus Florisil cartridge with 6 ml of acetone and 6 ml of 0.5% NH₄OH/acetone (v/v). The eluent was collected and concentrated to just dryness under a gentle stream of nitrogen gas. The dry residue was dissolved in 1 ml of methanol with supersonic extraction, and cleaned-up with Sep-Pak Plus NH₂. Free (unconjugated) estrogen was eluted with 5 ml of methanol, and conjugate estrogens were eluted with 6 ml of 0.5% NH₄OH/methanol (v/v). The collected eluent was blown down to dryness with a gentle stream of nitrogen. The dry residue was dissolved in 1 ml of acetonitrile/H₂O (6:4, v/v), which were then analyzed by LC/MS/MS. Operating conditions of the LC/MS/MS are presented in Table 1.

	Table 1. Ana	lytical Con	ditions of LC/MS/MS				
	Type of HPLC	Agilent 1100					
HPLC	Column	Agilent Zorbax Extend-C18, 2.1\phi \times 150mm, 40°C					
HPLC	Eluent	Acetonitrile 1mM NH4OH = 64, 0.14ml/min					
	Sample size	10μl					
	Type of MS/MS	TSQ API-2					
	Ionaization	AP-ESI, Negative					
	Collision gas	Argon					
		E1	269, 145 (50eV)				
	Measurment ion (collision energy)	E1-d2	271, 147 (50eV)				
		E2	271, 145 (45eV)				
		E2-d3	274, 145 (45eV)				
		EE2	295, 145 (45eV)				
		EE2-d4	299, 147 (45eV)				
MS/MS		E3	287, 171 (45eV)				
1013/1013		E3-d2	289, 173 (45eV)				
		E1-3S	349, 269 (35eV)				
		E2-S	351, 271 (35eV)				
		E2-S-d4	355, 275 (35eV)				
		E3-S	367, 287 (35eV)				
		E1-3G	445, 269 (35eV)				
		E2-G	447, 271 (35eV)				
		E3-G	463, 287 (35eV)				
		E2-S&G	527, 351 (35eV)				
		E2-diS	431, 351 (35eV)				

The field surveys were conducted at twenty WWTPs where influent and secondary effluent were collected for the analysis of estrogens and their conjugates (i.e., E1, E2, EE2, estriol (E3), estrone-3-sulfate (E1-S); β -estradiol 3-sulfate (E2-S); estriol 3-sulfate (E3-S); estrone β -D-glucuronide (E1-G); β -estradiol 17-(β -D)-glucuronide (E2-G); estriol 3-(β -D)-glucuronide (E3-G); β -estradiol 3-sulfate 17-glucuronide (E2-S&G); and estradiol 3,17-disulfate (E2-diS). The capacities of these WWTPs range from 12,000 to 680,000 m³/day. Thirteen of them apply a conventional activated sludge process. Three WWTPs employ an anaerobic-oxic activated sludge proceee (A/O process). Other WWTPs adapted various combined process: i.e., a conventional activated sludge process with rapid filtration; a conventional activated sludge process with rapid filtration and carbon adsorption; or an anaerobic-anoxic-oxic process (A2/O process) with rapid filtration and step aeration process. Grab samples were collected at WWTP sites. One gram of L-ascorbic acid was added to 1 litter of sample to prevent oxidation. All samples were collected in one-litter glass bottles, refrigerated, and transported to the laboratory within one day. Concentrations of estrogens and estrogen conjugates were measured by the method of Komori *et al.*,(2003).

Field survey of nonylphenol (NP) and its related substances

12 WWTPs were selected for the survey of nonylphenol and its related substances (i.e., NP, NPnEO (n=1-15), NPnEC (n=1-10)). All the WWTPs applied a conventional activated sludge process. The capacities of these WWTPs range from 17,300 to 168,000 m³/day. Grab samples were taken from the influent and the secondary effluent. One gram of L-ascorbic acid was added to 1 litter of sample to prevent oxidation. All samples were collected in one-litter glass bottles, refrigerated, and transported to the laboratory within one day. Concentrations of NP and NPEOs were measured by HPLC. HPLC was performed by the method of Komori *et al.*, (2002). Concentrations of NPnECs were measured by LC/MS/MS. LC/MS/MS was performed by the method of Yasojima *et al* (2002b).

RESULT AND DISCUSSION

Analytical Conditions for the LC/MS/MS Method

We examined the HPLC and MS/MS conditions for the optimal analysis of the estrogens and their conjugates. The LC column used was an Agilent Zorbax Extend-C18. In operating the MS/MS with electrospray ionization, better sensitivity was obtained for the estrogens and their conjugates as they were analyzed in a negative mode than in a positive mode. Table 1 lists MS/MS collision energies optimized for each compound. Fig. 3 shows the chromatograms of the standard solutions containing 100 μ g/l of each target compound and 40 μ g/l of its internal standard. Calibration curves were constructed for the quantification of the estrogens and their conjugates. A linear regression analysis was performed on the standard solution using the ratio of standard area to internal standard area as follows; E1-d2 for E1, E2-d3 for E2, EE2-d4 for EE2, E3-d2 for E3, and E2-S-d4 for E1-S, E2-S, E3-S, E1-G, E2-G, E3-G, E2-S&G and E2-diS. Linearity of the calibration curve obtained from the analysis of 0.5, 1.0, 2.0, 5.0 and 10 μ g/l of each analyte was high ($r^2 > 0.99$) for all the standard curves.

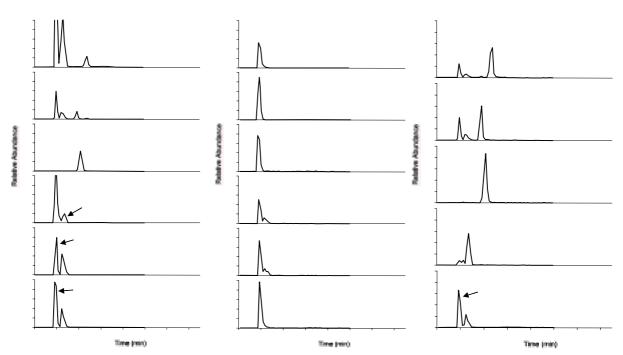


Figure 3 Chromatograms of E1, E2, EE2, E3, E1-S, E2-S, E3-S, E2-S&G, E2-diS, E1-G, E2-G, E3-G, E1-d2, E2-d3, EE2-d4, E3-d2 and E2-S-d4.

Detection Limits and Recovery Efficiencies of the LC/MS/MS Method

Detection Limit and Recovery Efficiency of the developed analytical method were examined using the standard solutions. The concentrations of the standard solutions, measured concentration (mean), standard deviation (σ), and detection limit for each compound are presented in Table 2. The detection limit was defined as three standard deviation (3σ) of the measurements divided by the concentration of the standard solution. When a wastewater sample required concentration more than 500 times, the detection limit for each estrogen and its conjugates was estimated (Table 2). The recovery efficiency of the analytical method was evaluated by spiking 20 ng of each compound to 500 ml of test samples; i.e., purified water, secondary settling tank effluent, and WWTP influent.

Table 2. Concentrations, Standard Deviations, and Detection limits (ng/l) for Selected Estrogens

	Concentration of standard solution	Measurement	σ	3σ	Detection limit
		(Average)	1.40	120	for sample
E1	500	340	140	420	0.8
E2	500	410	80	250	0.5
EE2	500	770	190	580	1.2
E3	500	580	240	710	1.4
E1-S	500	490	10	40	0.1
E2-S	500	480	40	120	0.2
E3-S	500	510	40	110	0.2
E1-G	500	820	220	670	1.3
E2-G	500	570	70	200	0.4
E3-G	500	51	90	260	0.5
E2-S&G	500	400	30	100	0.2
E2-diS	500	540	50	150	0.3

The recovery efficiencies of the estrogens with this method are presented in Table 3. The calculated recovery efficiencies for all the target compounds from purified water were higher than 75%. The recoveries from the secondary effluent and WWTP influent were higher than 94% for the free (unconjugated) estrogens (i.e., E1, E2, EE2, and E3), while they were less than 50% for the conjugated estrogens (i.e., E1-S, E2-S, E3-S, E1-G, E2-G, E3-G, E2-S&G, and E2-diS). Especially, the recoveries from the WWTP influent were calculated to be less than 15%.

Table 3. Recoveries (%) of estrogens from Purified water Samples through the Analytical Procedure

	Purified water	Secondray settling tank effluent	WWTP influent	
E1	100	103	110	
E2	106	100	104	
EE2	94	95	94	
E3	100	97	101	
E1-S	98	49	10	
E2-S	95	51	9.5	
E3-S	93	42	12	
E1-G	78	32	15	
E2-G	80	22	8.5	
E3-G	75	18	11	
E2-S&G	104	23	7.5	
E2-diS	102	81	6.5	

Field survey of estrogens and their conjugates

The measured concentrations of the target compounds in wastewater are presented in Table 4. In the WWTP influent, we found: 10 - 57 ng/l (median, 24 ng/l) of E1; ND - 21 ng/l (median, 5.7 ng/l) of E2; 27 - 220 ng/l (median, 110 ng/l) of E3; 12 - 170 ng/l (median, 42 ng/l) of E1-S; 26 - 410 ng/l (median, 110 ng/l) of E2-S; 6.5 - 79 ng/l (median, 22 ng/l) of E3-S; ND -88 ng/l (median, 11 ng/l) of E1-G; 5.3 - 100 ng/l (median, 18 ng/l) of E2-G; 4.1 - 73 ng/l (median, 22 ng/l) of E3-G; 0.8 - 38 ng/l (median, 5.5 ng/l) of E2-S&G; and 21 - 670 ng/l (median, 77 ng/l) of E2-diS. In the secondary effluent, we observed: ND - 180 ng/l (median, 12 ng/l) of E1; ND - 11 ng/l (median, ND) of E2; ND - 5.8 ng/l (median, 1.5 ng/l) of E3; 7.5 -34 ng/l (median, 13 ng/l) of E1-S; 27 - 94 ng/l (median, 52 ng/l) of E2-S; 37 - 160 ng/l (median, 69 ng/l) of E3-S; 34 - 140 ng/l (median, 74 ng/l) of E1-G; 47 - 210 ng/l (median, 91 ng/l) of E2-G; 37 - 150 ng/l (median, 72 ng/l) of E3-G; 3.7 - 17 ng/l (median, 8.9 ng/l) of E2-S&G; and 160 - 1500 ng/l (median, 360 ng/l) of E2-diS. EE2 was not detected in any of the samples analyzed (including WWTP influent and secondary effluent). The concentrations of E1, E2, and E3 were the same levels as those reported in the literature (Tanaka et al., 2003; MLIT, 2001b; Huang et al., 2001; Komori et al., 2001; Ternes et al., 1999a and Belfroid et al., 1999). Reductions of E2 and E3 (free, unconjugated estrogens) in the WWTPs were very good having nearly 100% and 98% reductions (calculated using median value), respectively. Reduction of E1 is 47%, which was considerably smaller than those of E2 and E3. This observation is consistent with Ternes et al. (1999b) who reported that the degradation rate of E1 is smaller than that of E2. Belfroid et al. (1999) reported that hormone-glucuronides exist generally below their detection limits in the effluent of WWTPs. However, the concentrations of the conjugated estrogens that we measured were higher than those of the unconjugated (free) estrogens in spite of the lower recovery ratios. The average concentrations of E1-S and E2-S (conjugated estrogens) were reduced in the WWTPs, but other conjugated estrogens (i.e., E3-S, E1-G, E2-G, E3-G, E2-S&G and E2-diS) were unchanged or increased. The removal efficiencies of E1-S and E2-S (conjugated estrogens) were approximately 68% and 51%, respectively.

Table 4. Concentrations of Selected Estrogens Detected in Wastewater Samples

from Twenty WWTPs (ng/l)

				- \ J /		
		min	25%	median	75%	max
	E1	10	17	24	29	57
	E2	ND (<0.5)	1.9	5.7	8.6	21
	EE2	ND (<1.2)	ND	ND	ND	ND
	E3	27	52	110	130	220
	E1-S	12	21	42	78	170
influent	E2-S	26	52	110	220	410
IIIIIUEIIL	E3-S	6.5	12	22	41	79
	E1-G	ND (<1.3)	5.2	11	24	88
	E2-G	5.3	12	18	31	100
	E3-G	4.1	11	22	38	73
	E2-S&G	0.8	2.2	5.5	12	38
	E2-diS	21	41	77	120	670
	E1	ND (<0.8)	3.1	12	46	180
	E2	ND (<0.5)	ND	ND	ND	11
	EE2	ND (<1.2)	ND	ND	ND	ND
	E3	ND (<1.4)	0.9	1.5	2	5.8
	E1-S	7.5	8.8	13	17	34
effluent	E2-S	27	44	52	56	94
emuem	E3-S	37	56	69	77	160
	E1-G	34	58	74	82	140
	E2-G	47	76	91	110	210
	E3-G	37	55	72	90	150
	E2-S&G	3.7	6.1	8.9	9.6	17
	E2-diS	160	240	360	510	1500

ND: Not detected (less than detection limit)

Field survey of nonylphenol (NP) and its related substances

The measured concentrations of the target compounds in wastewater are presented in. In the secondary effluent, we observed: $0.10\text{-}1.0\mu\text{g/L}$ (median, $0.20\mu\text{g/L}$) of NP. Table 5 and Figure 3. In the WWTP influent, we found: $0.50\text{-}20\mu\text{g/L}$ (median, $1.7\mu\text{g/L}$) of NP.

Regarding NP related substances, in the WWTP influent, NPEOs from NP1EO to NP15EO (mainly NP6EO-NP8EO) were detected but there were hardly any NPECs. In the secondary effluent, there were hardly any NPEO whose EO chain length is more than 5 and NPECs from NP1EC to NP4EC were detected. These results indicate that reduction of long-chain-NPEOs (EO chain length is more than 5) in STPs were very good but reduction of short-chain- NPEOs were small. Moreover, it was indicated that NPECs were produced in aerobic wastewater treatment process and degradation rate of long-chain-NPEC (EO chain length is more than 5) is large but degradation rate of short-chain NPEC is smaller.

In order to understand behavior of target compounds in wastewater treatment process, we focus on A STP of the 12 STPs as an example. The observed concentrations of target compounds in A STP are presented in Figure 5, Figure 6.

Regarding NPEOs, in the WWTP influent, NPEOs from NP1EO to NP14EO were detected and it is indicated that NPEOs were degraded easily in wastewater treatment because concentrations of NPEOs decreased drastically in the secondary effluent. It is unknown why values of NP4EO in both the influent and the secondary effluent were relatively large.

Regarding NPECs, in the WWTP influent, concentrations were below $0.1\mu g/L$, but concentrations of NPECs increased during wastewater treatment. Concentration of NP2EC was largest and NPEC from NP1EC to NP5EC were observed in the secondary effluent. One possible explanation about accumulation of NPECs in wastewater treatment process is difference between degradation rate of NPEO and NPEC. On the other hand, it is indicated

Table 5 Results of NP and its relates substances in WWTPs

		Influent				Secondary Effluent			
		Detection	min	mediar	max	Detection	min ´	mediar	max
		limit				limit			
Nonylphenol	NP	0.10	0.50	1.7	20	0.10	0.10	0.20	1.0
	NP1EO	0.04	0.82	2.1	17	0.04	0.04	0.26	0.49
	NP2EO	0.04	0.54	2.9	11	0.04	0.16	0.30	1.5
	NP3EO	0.06	1.2	4.5	14	0.06	0.07	0.14	0.81
	NP4EO	0.04	3.5	9.8	21	0.04	0.60	0.98	1.7
	NP5EO	0.05	2.0	7.3	23	0.05	N.D.	N.D.	0.08
	NP6EO	0.07	2.4	7.8	24	0.07	N.D.	N.D.	0.14
	NP7EO	0.05	2.4	7.9	23	0.05	N.D.	N.D.	0.14
Nonylphenol	NP8EO	0.08	2.8	7.5	24	0.08	N.D.	N.D.	N.D.
Ethoxylate	NP9EO	0.07	2.6	6.2	20	0.07	N.D.	N.D.	N.D.
	NP10EO		2.0	5.2	18	0.15	N.D.	N.D.	N.D.
	<u>NP11E0</u>	0.07	1.1	3.9	15	0.07	N.D.	N.D.	N.D.
	NP12EO	0.14	0.73	2.7	12	0.14	N.D.	N.D.	N.D.
	NP13EO		0.23	1.7	11	0.18	N.D.	N.D.	N.D.
	NP14E0	0.16	0.35	1.2	6.7	0.16	N.D.	N.D.	N.D.
	NP15EO	0.23	0.34	1.5	3.4	0.23	N.D.	N.D.	N.D.
	NP1EC	0.002	0.085	0.15	0.78	0.002	0.35	1.2	3.4
	NP2EC	0.002	0.11	0.36	4.7	0.002	1.1	3.5	10
	NP3EC	0.002	0.098	0.20	2.5	0.002	0.51	1.2	4.4
	NP4EC	0.002	0.086	0.14	0.99	0.002	0.13	0.52	2.5
Nonylphenoxy	NP5EC	0.002	0.088	0.14	0.88	0.002	0.065	0.24	1.2
Acetic Acid	NP6EC	0.002	0.11	0.14	0.48	0.002	0.009	0.049	0.62
7.000071010	NP7EC	0.002	0.054	0.13	0.50	0.002	0.006	0.019	0.31
	NP8EC	0.002	0.052	0.12	0.53	0.002	0.005	0.028	0.27
	NP9EC	0.002	0.055	0.11	0.46	0.002	N.D.	0.028	0.13
	NP10EC	0.002	0.057	0.11	0.45	0.002	N.D.	0.023	0.069
Unit a/L N.D.	an datact					-		•	

Unit g/l N.D. non-detection

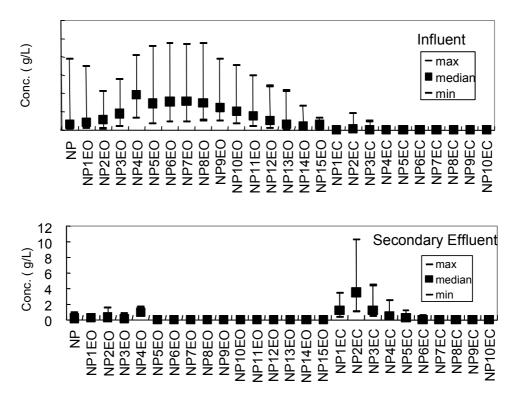
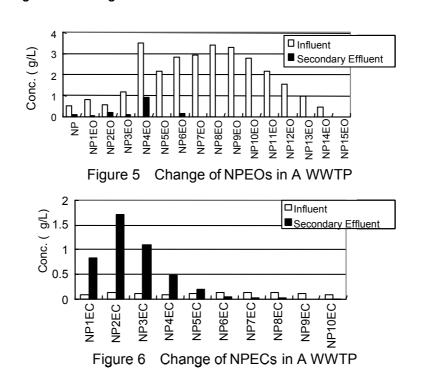


Figure 4. Change of NP and its related substances in WWTPs



that long-EO-chain NPEO changed to short-EO-chain NPEO and further to NPEC because long-EO-chain NPEC were not produced. Concentrations of NP in the secondary effluent were smaller than those in the influent. It was not unclear whether NP was biodegraded or removed by adsorption to sludge because concentrations on/in sludge were not measured.

CONCLUSIONS

- 1) LC/MS/MS method by Komori et al.,(2003) was applied to the wastewater samples collected from twenty WWTPs. The concentrations (median) of estrogens and their conjugates in the WWTP influent range from non-detection (ND) to as high as >100 ng/L. In the influent samples, the concentrations of E1, E2 and E3 were the same levels as those were previous reported. Belfroid *et al.* (1999) reported that hormone-glucuronides exist generally below their detection limits in effluent of WWTPs. However, the concentrations of conjugated estrogens that we measured were higher than those of free estrogens.
- 2) The reduction of the free estrogens in the WWTPs was very good with approximately 100% and 98% for E2 and E3, respectively, while removal efficiency for E1 (47%) was less significant than E2 and E3, suggesting that the degradation rate of E1 was smaller than that of E2 in the wastewater treatment processes. The concentrations of the conjugated estrogens (E1-S and E2-S) were declined in the WWTPs, while the levels of other conjugated estrogens (E3-S, E1-G, E2-G, E3-G, E2-S&G and E2-diS) were increased in the WWTPs.
- 3) HPLC method and LC/MS/MS method by Yasojima et al., (2002a, 2002b) were applied to the wastewater samoles from twelve WWTPs. The concentrations (median) of nonylpheol and its related substances in the WWTP influent range from non-detection (ND) to as high as >20 g/L. In the influent samples, the concentrations of NP were the same levels as those were previous reported.
- 4) The reduction of the long-EO-chain NPEOs in the WWTPs was very good with approximately 100%, respectively, while removal efficiency for short-EO-chain NPEOs was less significant than long-EO-chain NPEO, suggesting that the degradation rate of short-EO-chain NPEOs were smaller than those of long-EO-chain NPEO in the wastewater treatment processes. The concentrations of the NP were declined in the WWTPs, while the levels of short-EO-chain NPECs were increased in the WWTPs.

ACKNOWLEDGMENTS

The authors thank the local governments for kindly providing wastewater samples. We also thank the laboratory staff who prepared the samples for the HPLC and the LC/MS/MS analyses.

REFERENCES

Belfroid, A.C., Horst, A., Van der, Vethaak, A.D., Schafer, A.J., Rijs, G.B.J., Wegener, J. and Cofino, W.P. (1999) Analysis and occurrence of estrogenic hormones and their glucuronides in surface water and waste water in The Netherlands, The Science of the Total Environment, 225, 101-108.

Colborn, T., Dumanoski, D. and Myers, J.P. (1996) Our Stolen Future: Are We Threatening Our fertility, Intelligence, and Survival?-A scientific detective Story, Dutton.

Goda, T., Hirobe, M., Kobayashi, A., Fujimoto, S., Ike, M., Fujita, M., Okayasu, Y., Komori, K. and Tanaka, H. (2001) Development of ELISA for Detection of Estrogenic Hormones in Environment, CDROM of Proceedings of IWA 2nd World Water Congress, Berlin, Germany.

Huang, C.-H. and Sedlak, D.L. (2001) Analysis of estrogenic hormones in municipal wastewater effluent and surface water using enzyme linked immunosorbent assay and gas chromatography/tandem mass spectrometry, Environment Toxicology and Chemistry, 20(1), 133-139.

Japan Environment Agency, JEA, (1998) SPEED'98/JEA-Strategic Programs on Environmental Endocrine Disruptors'98, Japan Environment Agency.

Komori, K., Takahashi, A. and Tanaka, H. (2001) Detection of estrogens in wastewater by LC/MS/MS, CDROM of Proceedings of IWA 2nd World Water Congress, Berlin, Germany, (Paper No: P2063).

Komori, K., Yasojima, M., Tanaka, H., Sueoka, M., Ohiwa, T. and Kuno, M. (2002) Detection of estrogens in wastewater by LC/MS/MS, Proceedings of 5th Symposium of Japan Society on Water Environment, 186 (in Japanese).

Komori, K., Tanaka, H., Okayasu, Y., Yasojima, M. and Sato, C. (2003) Analysis and Occurrence of Estrogen in Wastewater in Japan, Proceedings of the 4th IWA Specialized Conference on Assessment and Control of Hazardous Substances in Water –ECOHAZARD 2003-, 45/1-45/8

Ministry of Land, Infrastructure and Transport, MLIT, (2001a) FY2000 Results of a Fact-finding Study of Endocrine Disruptors in Water Environments, Ministry of Land, Infrastructure and Transport (in Japanese).

Ministry of Land, Infrastructure, Transport, MLIT (2001b) Final Report on Countermeasures for Endocrine Disruptors in Sewerage Systems, Ministry of Land, Infrastructure, Transport (in Japanese).

Tanaka, H., Yakou, Y., Takahashi, A., Komori, K. and Okayasu, Y. (2001a). Evaluation of environmental estrogens in Japanese rivers, CDROM of Proceedings of WEFTEC'2001.

Tanaka, H., Yakou, Y., Takahashi, A., Higashitani, T., and Komori, K. (2001b). Comparison between estrogenicities estimated from DNA recombinant yeast assay and from chemical analyses of endocrine disruptors during sewage treatment. Wat.Sci. Tech., 43(2), 125-132.

Tanaka, H., Sato, C., Komori, K., Yakou, Y., Tamamoto, H., Miyamoto, N., and Higashitani T. (2003). Occurrence of endocrine disruptors in sewage and their behavior in sewage treatment plants in Japan, Environment Science, 10, 389-411.

Ternes, T.A., Stumpf, M., Mueller, J., Haberer, K., Wilken, R.D. and Servos, M. (1999a) Behavior and occurrence of estrogens in municipal sewage treatment plants-I. Investigations in Germany, Canada and Brazil, The science of the Total Environment, 225, 81-90.

Ternes, T.A., Kreckel, P. and Mueller, J. (1999b) Behavior and occurrence of estrogens in municipal sewage treatment plants-II. Aerobic batch experiments with activated sludge, The science of the Total Environment, 225, 91-99.

Yakou, Y., Takahashi, A., Higashitani, T., and Tanaka, H. (1999) Measurement of Estrogen-like Activity in Wastewater Using Recombinant Yeast, Environmental Engineering Research, 36, 199-208, Japan Society of Civil Engineering, (in Japanese).

Yasojima, M., Komori, K., Tanaka, H. (2002a). The Novel Analysis Method for Estrogens and Decomposition Products of Nonylphenol Ethoxylate in Wastewater, Proceedings of Water Korea 2002, B37-B40 (in Korea).

Yasojima, M., Komori, K., Tanaka, H. (2002b). The Analysis of Nonylphenoxyacetic Acids in Water of the Sewage Treatment Plant, Proceedings of 36th Congress of Japan Society on Water Environment, 526 (in Japanese)



Occurrence of endocrine disrupters in wastewater in Japan

Yuji Okayasu, Yutaka Suzuki, Koya KOMORI, and Makoto Yasojima

Water Quality Research Team,
Water Environment Research Group, Public
Works Research Institute (PWRI)

01/25

PWRI

MLIT Nation-wide Survey Briefing

- Ministry of Land Infrastructure Transport (MLIT) surveyed from FY 1998 to 2000
- 34 suspected endocrine disrupting chemical and related concerned chemicals (COC)
- 47 sewage treatment plants
- Investigated
 - Occurrence of COC in sewage and final discharge
 - Removal Efficiency in secondary and tertiary treatment
 - Occurrence of COC in sludge

03/25



PWRI Estrogen-like Activity Study

- Yeast estrogen screen (YES) reporter-gene assay provided by Brunel University group
- Secondary treatment can effectively reduce estrogen-like activity by 90%
- Among 34 COC, Estrogens & Nonylphenol seem key compounds from estrogenic significance

05/25

Background



- > Estrogens are deemed to be important substances that give estrogenic effects on fish in the water environment
- ➤ Most estrogens are hydroxylated and conjugated to glucronides, sulfates, and acetates in excretion
- ➤ Very few analytical methods are capable of analyzing conjugated estrogens (Ternes et al., Belfroid et al.)
- Relatively little work has been directed toward investigating impacts and occurrence of estrogens in water environment

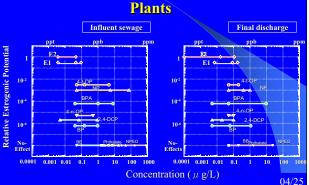
Outline of the presentation



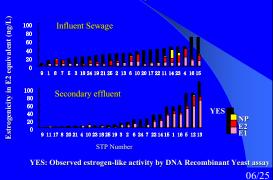
- ✓ Introduction
- ✓ Background
- **✓** Objective
- ✓ Method
- ✓ Result
- ✓ Conclusion

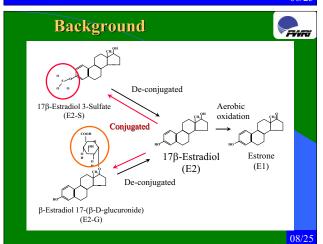
02/25

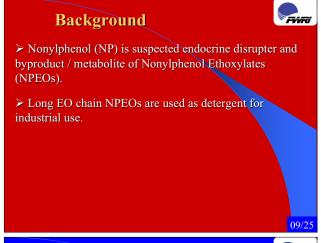
Occurrence & Estrogenic Potential of COCs Surveyed in Sewage Treatment Plants



Gap between YES and Estrogenic Contribution of Surveyed Chemicals





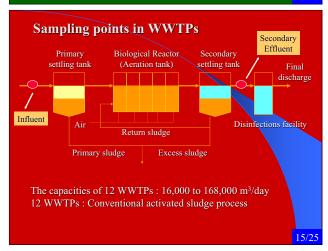


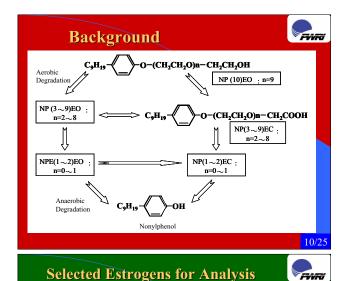
PVVRI **Objectives**

> Study occurrence of estrogens and their conjugates in wastewater and evaluate of estrogens and their conjugates in WWTP performance

> Study occurrence of NP and its related substances (NPEOs and NPECs) in wastewater and evaluate of them in WWTP performance

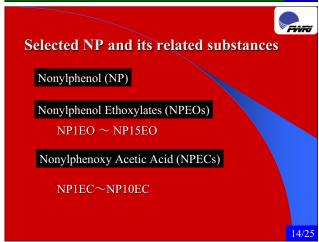
Sampling points in WWTPs Secondary Effluent Biological Reactor settling tank (Aeration tank) settling tank Final Disinfections facility Return sludge Primary sludge Excess sludge The capacities of 20 WWTPs: 12,000 to 680,000 m³/day 13 WWTPs: Conventional activated sludge process Other WWTPs: A/O process, A2/O process and so on 13/25





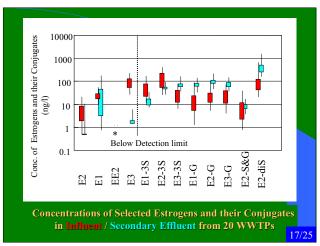


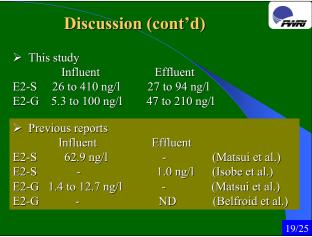
PVVFi





11/25





Conclusions

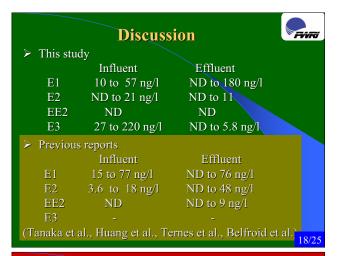
- > The reduction of the free estrogens in the WWTPs was very good with approximately 100% and 98% for E2 and E3, respectively, while removal efficiency for E1 (47%) was less significant than E2 and E3
- ➤ The concentrations of the conjugated estrogens (E1-S and E2-S) were decreased in the WWTPs, while the levels of the other conjugated estrogens (E3-S, E1-G, E2-G, E3-G, E2-S&G and E2-diS) were unchanged or rather increased in the WWTPs

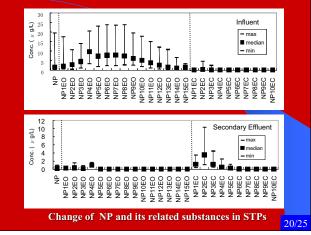


PVRi

Future challenge

- > Fate of estrogens and their conjugates in wastewater treatment plant and the water environment and their effect on aquatic organisms in the water environment.
- Fate of nonylphenol and its related substances in wastewater treatment plant and the water environment and their effect on aquatic organisms in the water environment.





PVRI

Conclusions (cont'd)

- ➤ The concentrations of the nonylphenol (NP) and long EO chain nonylphenol ethoxylates (NPEOs) were decreased in the wastewater treatment process.
- ➤ little short EO chain NPEOs and considerable short EO chain nonylphenoxy acetic acid (NPECs) remained in final effluent.

22/25



Acknowledgement

- ➤ The authors thank the local governments for kindly providing wastewater samples.
- ➤ We also thank the laboratory staff who prepared the sample for analysis.

24/25

